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cloud) Ionic Bond. Chemical bonding that results from the electrical attraction between large numbers of cations and anions. (When metal gives electron ...

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greater than 1.7 and is a transfer of electrons. The Lower Electronegativity transfers to the higher electronegativity)  
t. Ionic Bond. Ions pack in a crystal lattice structure. Lattice Energy. Cation {+} and anion {-} create \_\_\_\_\_

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REVIEW Chemical Bonding SECTION 4

SHORT ANSWER Answer the following

questions in the space provided. 1. b In

metals, the valence electrons are

considered to be (a) attached to

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particular positive ions. (c) immobile.

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18.  $3s^2 3p^4$  19. ...

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SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Use the concept of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the

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CHAPTER 6 Chemical Bonding SECTION

1 Introduction to Chemical Bonding

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OBJECTIVES 1. Define Chemical bond. 2. Explain why most atoms form chemical bonds. 3. Describe ionic and covalent bonding.. 4.

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SECTION 3 SHORT ANSWER Answer the following questions in the space provided 1 \_\_\_\_ The notation for sodium chloride, NaCl, stands for one (a) formula unit (c) crystal (b) molecule (d) atom 2 \_\_\_\_ In a crystal of an ionic compound, each ...

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CHAPTER 12 REVIEW Solutions SECTION  
2 SHORT ANSWER Answer the following  
questions in the space provided. 1. The  
following are statements about the  
dissolving process. Explain each one at  
the molecular level. a. Increasing the  
pressure of a solute gas above a liquid  
solution increases the solubility of the

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